

1-[4-[Bis(4-fluorophenyl)methyl]-piperazin-1-yl]ethanone

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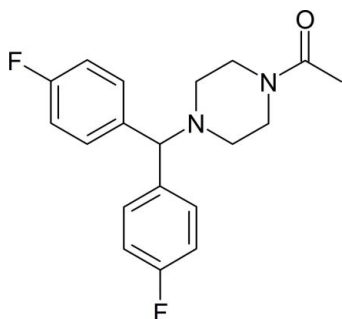
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Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.054; wR factor = 0.155; data-to-parameter ratio = 15.1.

In the title compound, $\text{C}_{19}\text{H}_{20}\text{F}_2\text{N}_2\text{O}$, the six-membered piperazine group adopts a slightly distorted chair conformation. The dihedral angle between the mean planes of the two benzene rings is 73.4 (6)°. The mean plane of the ethanone group is twisted from the mean planes of the two benzene rings by 66.7 (8) and 86.2 (6)°. In the crystal, $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{F}$ interactions link the molecules, forming a three-dimensional structure.

Related literature

For the biological activity of piperazines, see: Bogatcheva *et al.* (2006); Brockunier *et al.* (2004). For a review of pharmacological and toxicological information for piperazine derivatives, see: Elliott (2011). For related structures, see: Betz *et al.* (2011a,b); Dai *et al.* (2012); Dayananda *et al.* (2012a,b); Zhong *et al.* (2011). For puckering parameters, see: Cremer & Pople (1975). For reference bond-length data, see Allen *et al.* (1987).



Experimental

Crystal data

$\text{C}_{19}\text{H}_{20}\text{F}_2\text{N}_2\text{O}$
 $M_r = 330.37$
 Monoclinic, $P2_1/n$
 $a = 10.1701$ (5) Å
 $b = 16.5521$ (5) Å
 $c = 11.1690$ (5) Å
 $\beta = 114.690$ (5)°
 $V = 1708.27$ (14) Å³

$Z = 4$
 Cu $K\alpha$ radiation
 $\mu = 0.79$ mm⁻¹

$T = 173$ K
 $0.48 \times 0.32 \times 0.22$ mm

Data collection

Oxford Xcalibur Eos Gemini diffractometer
 Absorption correction: multi-scan (*CrysAlis RED*; Oxford Diffraction, 2010)
 $T_{\min} = 0.802$, $T_{\max} = 1.000$
 10536 measured reflections
 3297 independent reflections
 2809 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.026$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.054$
 $wR(F^2) = 0.155$
 $S = 1.04$
 3297 reflections
 218 parameters
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.41$ e Å⁻³
 $\Delta\rho_{\min} = -0.34$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C13}-\text{H13}\cdots\text{O1}^i$	0.93	2.46	3.371 (2)	167
$\text{C15}-\text{H15}\cdots\text{O1}^i$	0.93	2.55	3.351 (3)	145
$\text{C18}-\text{H18}\cdots\text{F2}^{ii}$	0.93	2.54	3.319 (3)	142

Symmetry codes: (i) $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-x + 2, -y + 2, -z + 1$.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2010); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis RED* (Oxford Diffraction, 2010); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZJ2086).

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supporting information

Acta Cryst. (2012). E68, o2237 [https://doi.org/10.1107/S1600536812028097]

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S1. Comment

4,4'-Difluorobenzhydryl piperazine is an intermediate for the preparation of flunarizine which is a calcium channel blocker. Piperazines are among the most important building blocks in today's drug discovery and are found in biologically active compounds across a number of different therapeutic areas (Brockunier *et al.*, 2004; Bogatcheva *et al.*, 2006). A review on the current pharmacological and toxicological information for piperazine derivatives is described (Elliott, 2011).

The crystal structures of 4-[bis(4-fluorophenyl)methyl]piperazin-1-ium 2-(2-phenylethyl) benzoate (Betz *et al.*, 2011*a*), 4-[bis(4-fluorophenyl)methyl]piperazin-1-ium picrate (Betz *et al.*, 2011*b*), (*E*)-1-{4-[bis(4-fluorophenyl)methyl]-piperazin-1-yl}-3-(4-ethoxyphenyl) prop-2-en-1-one (Zhong *et al.*, 2011), 4-[bis(4-fluorophenyl) methyl]piperazin-1-ium bis(trichloroacetate) 0.4-hydrate (Dayananda *et al.*, 2012*a*), 4-[bis(4-fluorophenyl)methyl] piperazin-1-ium 2-hydroxybenzoate 2-hydroxybenzoic acid monosolvate (Dayananda *et al.*, 2012*b*) and 1-[bis(4-fluorophenyl) methyl]-4-[2-(2-methylphenoxy)ethyl]piperazine (Dai *et al.*, 2012) have been reported. In the course of our studies on the salts of piperazines and in view of the importance of piperazines, this paper reports the crystal and molecular structure of the title compound, C₁₉H₂₀F₂N₂O, (I), which was accidentally obtained by the reaction of 4,4'-difluorobenzhydryl piperazine and acetyl salicylic acid.

In the asymmetric unit of the title compound, (I), the 6-membered piperazine group (N1/C3/C4/N2/C5/C6) adopts a slightly distorted chair conformation with puckering parameters *Q*, θ and φ of 0.568 (9) Å, 172.2 (7)°, and 350.979 (8)° (Cremer & Pople, 1975), respectively (Fig. 1). For an ideal chair θ has a value of 0 or 180°. Bond lengths are in normal ranges (Allen *et al.*, 1987). The dihedral angle between the mean planes of the two benzene rings is 73.4 (6)°. The mean plane of the ethanone group (C1/C2/O1/N1) is twisted from the mean planes of the two benzene rings by 66.7 (8)° and 86.2 (6)°. Weak C—H...O and C—H...F intermolecular interactions (Table 1) are observed providing increased stability with crystal packing (Fig. 2).

S2. Experimental

4,4'-Difluorobenzhydryl piperazine was obtained from R. L. Fine Chem., Bengaluru, India. 4,4'-Difluorobenzhydryl piperazine (2.88 g, 0.01 mol) was dissolved in 10 ml of absolute ethanol and acetylsalicylic acid (1.81 g, 0.01 mol) was also dissolved in 10 ml of absolute ethanol. Both the solutions were mixed and stirred in a beaker at 333 K for 30 min. The mixture was kept aside for a day at room temperature. The compound formed was filtered and dried in a vacuum desiccator over phosphorous pentoxide. The compound was recrystallized from a mixture of toluene and dimethyl formamide by slow evaporation (m.p. 418–423 K).

S3. Refinement

All of the H atoms were placed in their calculated positions and then refined using the riding model with C—H lengths of 0.93, 0.98 (CH) or 0.96 Å (CH₃). The isotropic displacement parameters for these atoms were set to 1.19 to 1.20 (CH), 1.19 to 1.20 (CH₂) or 1.50 (CH₃) times U_{eq} of the parent atom.

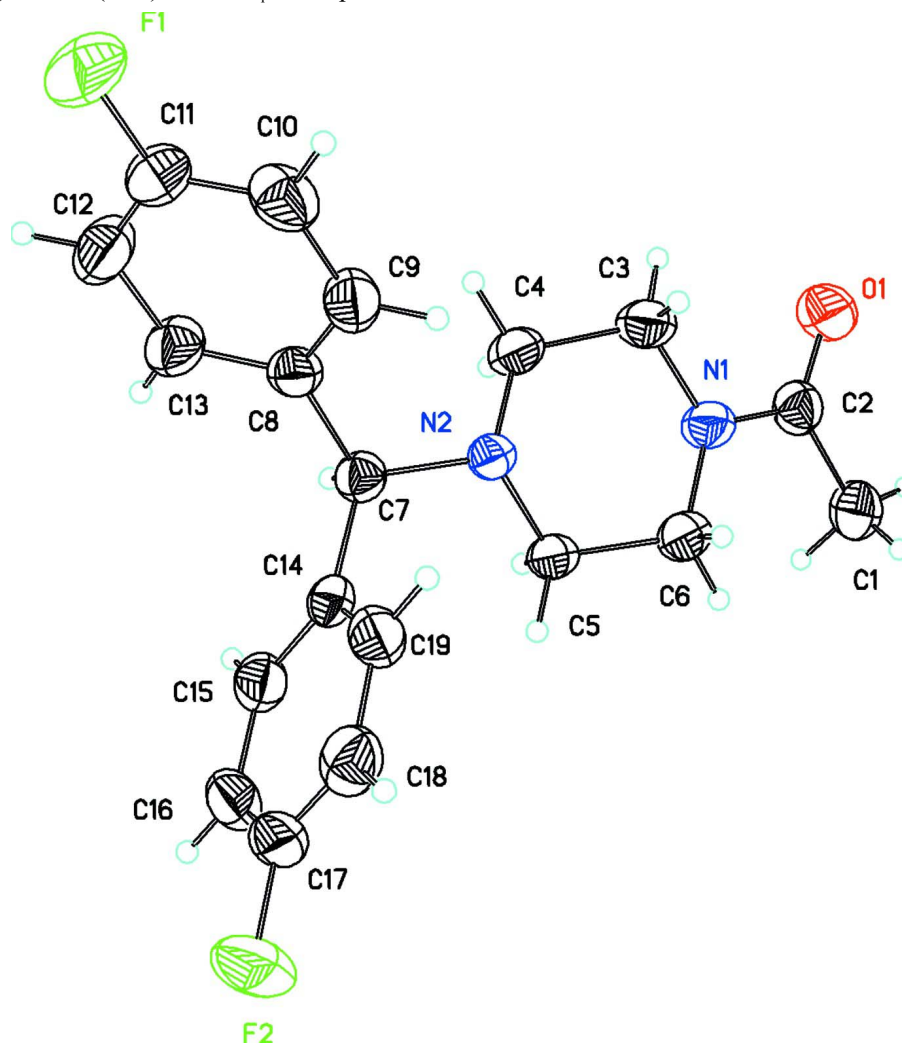


Figure 1

Molecular structure of the title compound showing the atom labeling scheme and 50% probability displacement ellipsoids.

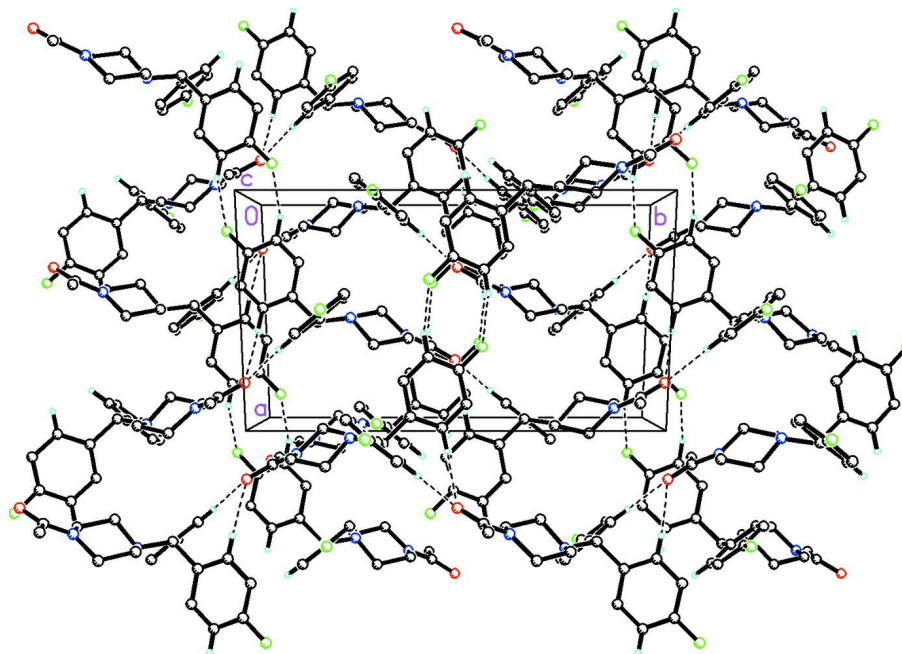


Figure 2

Packing diagram of the title compound viewed along the c axis. Dashed lines indicate weak C—H...O and C—H...F intermolecular interactions. The remaining H atoms have been removed for clarity.

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Crystal data

$C_{19}H_{20}F_2N_2O$

$M_r = 330.37$

Monoclinic, $P2_1/n$

Hall symbol: -P 2yn

$a = 10.1701(5) \text{ \AA}$

$b = 16.5521(5) \text{ \AA}$

$c = 11.1690(5) \text{ \AA}$

$\beta = 114.690(5)^\circ$

$V = 1708.27(14) \text{ \AA}^3$

$Z = 4$

$F(000) = 696$

$D_x = 1.285 \text{ Mg m}^{-3}$

Cu $K\alpha$ radiation, $\lambda = 1.54178 \text{ \AA}$

Cell parameters from 4103 reflections

$\theta = 4.4\text{--}71.2^\circ$

$\mu = 0.79 \text{ mm}^{-1}$

$T = 173 \text{ K}$

Chunk, colourless

$0.48 \times 0.32 \times 0.22 \text{ mm}$

Data collection

Oxford Xcalibur Eos Gemini

diffractometer

Radiation source: Enhance (Cu) X-ray Source

Graphite monochromator

Detector resolution: $16.1500 \text{ pixels mm}^{-1}$

ω scans

Absorption correction: multi-scan

(*CrysAlis RED*; Oxford Diffraction, 2010)

$T_{\min} = 0.802$, $T_{\max} = 1.000$

10536 measured reflections

3297 independent reflections

2809 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.026$

$\theta_{\max} = 71.4^\circ$, $\theta_{\min} = 5.0^\circ$

$h = -12 \rightarrow 12$

$k = -12 \rightarrow 20$

$l = -13 \rightarrow 13$

*Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.054$ $wR(F^2) = 0.155$ $S = 1.04$

3297 reflections

218 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0781P)^2 + 0.7598P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} < 0.001$ $\Delta\rho_{\max} = 0.41 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.34 \text{ e } \text{\AA}^{-3}$ *Special details*

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
F1	0.5160 (2)	0.80482 (10)	−0.24955 (14)	0.0811 (5)
F2	0.86017 (18)	1.06144 (9)	0.55722 (15)	0.0788 (5)
O1	0.29530 (17)	0.50309 (9)	0.41107 (15)	0.0534 (4)
N1	0.40533 (18)	0.62266 (10)	0.42534 (15)	0.0410 (4)
N2	0.48228 (15)	0.75448 (8)	0.30161 (13)	0.0304 (3)
C1	0.3741 (3)	0.57161 (13)	0.6169 (2)	0.0520 (5)
H1A	0.3212	0.5298	0.6371	0.078*
H1B	0.3382	0.6234	0.6280	0.078*
H1C	0.4749	0.5674	0.6752	0.078*
C2	0.3555 (2)	0.56254 (11)	0.47665 (19)	0.0395 (4)
C3	0.3933 (2)	0.61611 (12)	0.29096 (19)	0.0465 (5)
H3A	0.3142	0.5801	0.2406	0.056*
H3B	0.4819	0.5935	0.2922	0.056*
C4	0.3666 (2)	0.69801 (12)	0.22666 (17)	0.0406 (4)
H4A	0.3605	0.6930	0.1380	0.049*
H4B	0.2749	0.7189	0.2206	0.049*
C5	0.4817 (2)	0.76336 (11)	0.43201 (17)	0.0363 (4)
H5A	0.3886	0.7842	0.4222	0.044*
H5B	0.5555	0.8019	0.4835	0.044*
C6	0.5101 (2)	0.68390 (12)	0.50348 (18)	0.0426 (5)
H6A	0.6072	0.6658	0.5213	0.051*
H6B	0.5038	0.6909	0.5872	0.051*
C7	0.46380 (18)	0.83174 (11)	0.23205 (16)	0.0333 (4)
H7	0.3663	0.8519	0.2120	0.040*
C8	0.47860 (19)	0.82157 (11)	0.10209 (17)	0.0356 (4)

C9	0.5661 (2)	0.76331 (12)	0.0844 (2)	0.0430 (4)
H9	0.6165	0.7276	0.1526	0.052*
C10	0.5799 (2)	0.75726 (12)	−0.0344 (2)	0.0508 (5)
H10	0.6388	0.7181	−0.0465	0.061*
C11	0.5044 (3)	0.81058 (13)	−0.1324 (2)	0.0522 (5)
C12	0.4182 (3)	0.86953 (15)	−0.1176 (2)	0.0576 (6)
H12	0.3689	0.9055	−0.1856	0.069*
C13	0.4060 (2)	0.87437 (13)	0.00127 (19)	0.0472 (5)
H13	0.3477	0.9141	0.0129	0.057*
C14	0.5726 (2)	0.89431 (11)	0.31739 (16)	0.0349 (4)
C15	0.5244 (2)	0.96762 (12)	0.34201 (19)	0.0426 (4)
H15	0.4259	0.9791	0.3044	0.051*
C16	0.6209 (3)	1.02416 (12)	0.4218 (2)	0.0522 (5)
H16	0.5883	1.0736	0.4384	0.063*
C17	0.7645 (3)	1.00628 (13)	0.4757 (2)	0.0507 (5)
C18	0.8181 (2)	0.93527 (13)	0.4525 (2)	0.0496 (5)
H18	0.9171	0.9251	0.4897	0.060*
C19	0.7207 (2)	0.87896 (12)	0.37212 (19)	0.0419 (4)
H19	0.7546	0.8303	0.3544	0.050*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.1304 (14)	0.0839 (11)	0.0563 (9)	0.0159 (10)	0.0658 (9)	0.0004 (7)
F2	0.0951 (11)	0.0602 (9)	0.0724 (10)	−0.0414 (8)	0.0264 (8)	−0.0204 (7)
O1	0.0659 (9)	0.0428 (8)	0.0519 (8)	−0.0214 (7)	0.0250 (7)	−0.0029 (6)
N1	0.0530 (9)	0.0380 (8)	0.0314 (8)	−0.0133 (7)	0.0170 (7)	−0.0030 (6)
N2	0.0345 (7)	0.0299 (7)	0.0268 (7)	−0.0034 (6)	0.0128 (6)	−0.0012 (5)
C1	0.0722 (14)	0.0438 (11)	0.0454 (11)	−0.0056 (10)	0.0300 (10)	0.0073 (9)
C2	0.0417 (9)	0.0368 (10)	0.0405 (10)	−0.0024 (8)	0.0177 (8)	0.0035 (8)
C3	0.0651 (12)	0.0400 (10)	0.0374 (10)	−0.0168 (9)	0.0245 (9)	−0.0084 (8)
C4	0.0477 (10)	0.0448 (10)	0.0278 (8)	−0.0129 (8)	0.0143 (8)	−0.0043 (7)
C5	0.0448 (10)	0.0348 (9)	0.0278 (9)	−0.0085 (7)	0.0137 (7)	−0.0037 (7)
C6	0.0516 (11)	0.0410 (10)	0.0305 (9)	−0.0107 (8)	0.0126 (8)	−0.0011 (7)
C7	0.0355 (8)	0.0342 (9)	0.0315 (9)	0.0023 (7)	0.0152 (7)	0.0004 (7)
C8	0.0410 (9)	0.0347 (9)	0.0334 (9)	−0.0048 (7)	0.0181 (7)	−0.0038 (7)
C9	0.0492 (11)	0.0379 (10)	0.0447 (11)	0.0022 (8)	0.0224 (9)	0.0050 (8)
C10	0.0624 (13)	0.0402 (11)	0.0652 (14)	0.0035 (9)	0.0420 (11)	−0.0045 (9)
C11	0.0776 (14)	0.0523 (12)	0.0429 (11)	0.0011 (11)	0.0411 (11)	−0.0009 (9)
C12	0.0765 (15)	0.0617 (14)	0.0387 (11)	0.0184 (12)	0.0281 (11)	0.0106 (10)
C13	0.0594 (12)	0.0488 (11)	0.0398 (10)	0.0111 (9)	0.0271 (9)	0.0023 (8)
C14	0.0449 (9)	0.0345 (9)	0.0265 (8)	−0.0055 (7)	0.0161 (7)	0.0033 (7)
C15	0.0501 (11)	0.0404 (10)	0.0392 (10)	0.0029 (8)	0.0206 (8)	0.0048 (8)
C16	0.0821 (16)	0.0304 (10)	0.0503 (12)	−0.0004 (10)	0.0338 (11)	−0.0008 (8)
C17	0.0689 (14)	0.0413 (11)	0.0399 (10)	−0.0249 (10)	0.0207 (10)	−0.0024 (8)
C18	0.0448 (10)	0.0541 (12)	0.0472 (11)	−0.0080 (9)	0.0165 (9)	0.0082 (9)
C19	0.0500 (11)	0.0349 (10)	0.0442 (10)	0.0001 (8)	0.0232 (9)	0.0015 (8)

Geometric parameters (Å, °)

F1—C11	1.364 (2)	C7—C14	1.524 (2)
F2—C17	1.367 (2)	C7—C8	1.530 (2)
O1—C2	1.227 (2)	C7—H7	0.9800
N1—C2	1.349 (2)	C8—C13	1.372 (3)
N1—C3	1.458 (2)	C8—C9	1.381 (3)
N1—C6	1.465 (2)	C9—C10	1.396 (3)
N2—C4	1.462 (2)	C9—H9	0.9300
N2—C5	1.466 (2)	C10—C11	1.365 (3)
N2—C7	1.467 (2)	C10—H10	0.9300
C1—C2	1.505 (3)	C11—C12	1.368 (3)
C1—H1A	0.9600	C12—C13	1.386 (3)
C1—H1B	0.9600	C12—H12	0.9300
C1—H1C	0.9600	C13—H13	0.9300
C3—C4	1.505 (3)	C14—C15	1.378 (3)
C3—H3A	0.9700	C14—C19	1.393 (3)
C3—H3B	0.9700	C15—C16	1.379 (3)
C4—H4A	0.9700	C15—H15	0.9300
C4—H4B	0.9700	C16—C17	1.359 (3)
C5—C6	1.503 (3)	C16—H16	0.9300
C5—H5A	0.9700	C17—C18	1.365 (3)
C5—H5B	0.9700	C18—C19	1.383 (3)
C6—H6A	0.9700	C18—H18	0.9300
C6—H6B	0.9700	C19—H19	0.9300
C2—N1—C3	119.85 (15)	C14—C7—C8	109.55 (14)
C2—N1—C6	124.48 (15)	N2—C7—H7	108.2
C3—N1—C6	113.10 (15)	C14—C7—H7	108.2
C4—N2—C5	107.26 (13)	C8—C7—H7	108.2
C4—N2—C7	111.08 (13)	C13—C8—C9	119.04 (17)
C5—N2—C7	112.70 (13)	C13—C8—C7	118.44 (16)
C2—C1—H1A	109.5	C9—C8—C7	122.48 (16)
C2—C1—H1B	109.5	C8—C9—C10	120.84 (18)
H1A—C1—H1B	109.5	C8—C9—H9	119.6
C2—C1—H1C	109.5	C10—C9—H9	119.6
H1A—C1—H1C	109.5	C11—C10—C9	118.04 (18)
H1B—C1—H1C	109.5	C11—C10—H10	121.0
O1—C2—N1	121.36 (17)	C9—C10—H10	121.0
O1—C2—C1	121.08 (17)	F1—C11—C10	118.60 (19)
N1—C2—C1	117.55 (16)	F1—C11—C12	118.8 (2)
N1—C3—C4	110.18 (16)	C10—C11—C12	122.56 (19)
N1—C3—H3A	109.6	C11—C12—C13	118.4 (2)
C4—C3—H3A	109.6	C11—C12—H12	120.8
N1—C3—H3B	109.6	C13—C12—H12	120.8
C4—C3—H3B	109.6	C8—C13—C12	121.12 (19)
H3A—C3—H3B	108.1	C8—C13—H13	119.4
N2—C4—C3	111.16 (15)	C12—C13—H13	119.4

N2—C4—H4A	109.4	C15—C14—C19	118.84 (17)
C3—C4—H4A	109.4	C15—C14—C7	119.82 (17)
N2—C4—H4B	109.4	C19—C14—C7	121.33 (16)
C3—C4—H4B	109.4	C14—C15—C16	120.7 (2)
H4A—C4—H4B	108.0	C14—C15—H15	119.7
N2—C5—C6	111.14 (15)	C16—C15—H15	119.7
N2—C5—H5A	109.4	C17—C16—C15	118.8 (2)
C6—C5—H5A	109.4	C17—C16—H16	120.6
N2—C5—H5B	109.4	C15—C16—H16	120.6
C6—C5—H5B	109.4	C16—C17—C18	122.94 (19)
H5A—C5—H5B	108.0	C16—C17—F2	118.9 (2)
N1—C6—C5	111.02 (15)	C18—C17—F2	118.1 (2)
N1—C6—H6A	109.4	C17—C18—C19	118.0 (2)
C5—C6—H6A	109.4	C17—C18—H18	121.0
N1—C6—H6B	109.4	C19—C18—H18	121.0
C5—C6—H6B	109.4	C18—C19—C14	120.79 (19)
H6A—C6—H6B	108.0	C18—C19—H19	119.6
N2—C7—C14	111.23 (13)	C14—C19—H19	119.6
N2—C7—C8	111.40 (14)		
C3—N1—C2—O1	−2.5 (3)	C7—C8—C9—C10	−178.25 (18)
C6—N1—C2—O1	−163.00 (19)	C8—C9—C10—C11	0.0 (3)
C3—N1—C2—C1	178.49 (18)	C9—C10—C11—F1	−179.4 (2)
C6—N1—C2—C1	17.9 (3)	C9—C10—C11—C12	0.8 (4)
C2—N1—C3—C4	145.48 (18)	F1—C11—C12—C13	179.4 (2)
C6—N1—C3—C4	−51.9 (2)	C10—C11—C12—C13	−0.8 (4)
C5—N2—C4—C3	−62.16 (19)	C9—C8—C13—C12	0.6 (3)
C7—N2—C4—C3	174.28 (15)	C7—C8—C13—C12	178.3 (2)
N1—C3—C4—N2	58.1 (2)	C11—C12—C13—C8	0.1 (4)
C4—N2—C5—C6	60.94 (19)	N2—C7—C14—C15	−124.60 (17)
C7—N2—C5—C6	−176.50 (14)	C8—C7—C14—C15	111.80 (18)
C2—N1—C6—C5	−147.07 (18)	N2—C7—C14—C19	55.1 (2)
C3—N1—C6—C5	51.2 (2)	C8—C7—C14—C19	−68.5 (2)
N2—C5—C6—N1	−56.0 (2)	C19—C14—C15—C16	−1.4 (3)
C4—N2—C7—C14	172.68 (14)	C7—C14—C15—C16	178.28 (17)
C5—N2—C7—C14	52.28 (18)	C14—C15—C16—C17	0.0 (3)
C4—N2—C7—C8	−64.78 (18)	C15—C16—C17—C18	1.3 (3)
C5—N2—C7—C8	174.82 (14)	C15—C16—C17—F2	−178.55 (18)
N2—C7—C8—C13	151.90 (17)	C16—C17—C18—C19	−1.1 (3)
C14—C7—C8—C13	−84.6 (2)	F2—C17—C18—C19	178.75 (18)
N2—C7—C8—C9	−30.5 (2)	C17—C18—C19—C14	−0.4 (3)
C14—C7—C8—C9	93.0 (2)	C15—C14—C19—C18	1.6 (3)
C13—C8—C9—C10	−0.7 (3)	C7—C14—C19—C18	−178.09 (16)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C13—H13 \cdots O1 ⁱ	0.93	2.46	3.371 (2)	167

C15—H15 \cdots O1 ⁱ	0.93	2.55	3.351 (3)	145
C18—H18 \cdots F2 ⁱⁱ	0.93	2.54	3.319 (3)	142

Symmetry codes: (i) $-x+1/2, y+1/2, -z+1/2$; (ii) $-x+2, -y+2, -z+1$.